

D. MONOGRAPHS

Acetic Acid, Glacial



$\text{C}_2\text{H}_4\text{O}_2$

Mol. Wt. 60.05

Acetic acid

[64-19-7]

Content Glacial Acetic Acid contains not less than 99.0% of acetic acid ($\text{C}_2\text{H}_4\text{O}_2$)

Description Glacial Acetic Acid occurs as colorless or white crystalline lumps or is a colorless and clear liquid. It has a characteristic pungent odor.

Identification (1) Glacial Acetic Acid solution (1 4) is acidic.

(2) Glacial Acetic Acid solution (1 4) responds to all tests for Acetate as described in the Qualitative Tests.

Purity (1) Congeeing point Not less than 14.5 .

(2) Heavy metals Not more than 10 $\mu\text{g/g}$ as Pb (2.0 g, Method 1, control solution Lead Standard Solution 2.0 ml).

(3) Arsenic Not more than 4.0 $\mu\text{g/g}$ as As_2O_3 (0.5 0g, Method 1, Apparatus B).

(4) Readily oxidizable substances Weigh 2.0 g of Glacial Acetic Acid, dissolve in 10 ml of water, and add 0.10 ml of 0.2 mol/l potassium permanganate. The pink color of the solution does not disappear within 30 minutes.

(5) Residue on evaporation Not more than 0.010%.

Weigh 20.0 g of Glacial Acetic Acid, evaporate, dry at 100 for 2 hours, and weigh the residue.

Assay Weigh accurately about 1 g of Glacial Acetic Acid, and add 40 ml of water. Titrate with 1 mol/l sodium hydroxide (indicator: 2 drops of phenolphthalein TS).

1 ml of 1 mol/l sodium hydroxide = 60.05 mg of $\text{C}_2\text{H}_4\text{O}_2$